$(+)$ -Hupeol, a Possible Non-basic Metabolite of the Lupine Alkaloid $(-)$ -Cytisine in Chinese Maackia hupehensist

Yong-hong Wang,^a Hajime Kubo,^a Kimio Higashiyama,^a Hideaki Komiya, a^a Jia-Shi Li^b and Shigeru Ohmiya^{*a}

^a Department of Synthetic Organic Chemistry, Institute of Medicinal Chemistry, Hoshi University, Ebara 2-4-41, Shinagawa-ku, Tokyo 142, Japan ^bBeijing University of Traditional Chinese Medicine, 11, Beisan Huan Dong Ave, Beijing 100029, China

J. Chem. Research (S), 1998, 196±197\$

A novel product, named $(+)$ -hupeol (1), which is regarded as an intermediate in the metabolism of the lupine alkaloids to non-basic constituents, has been isolated from Chinese Maackia hupehensis.

Japanese Maackia species (Leguminosae) are a group of plants which are interesting from the viewpoints of chemotaxonomy and biosynthesis because they accumulate unusual lupine alkaloids containing a pyrrolizidine or indolizidine ring such as maackiamine¹, tashiromine² and camoensidine³ together with common lupine alkaloids having a piperidine or quinolizidine ring. In the course of our studies on lupine alkaloids in Maackia plants, we isolated a novel constituent from M . hupehensis native to China. In this paper, we report the chemical characterization of the new constituent, named $(+)$ -hupeol (1) , and its biogenetic relationship with the typical lupine alkaloid $(-)$ cytisine (2), which is a main alkaloid (25% of the total base) of this plant.

The basic fraction (5.4 g) obtained from a 75% MeOH extract of the dry branches (1.2 kg) of *M. hupehensis*, collected in Jiang Xi province, China, in May, was subjected to repeated column chromatography on silica gel to yield $(+)$ -hupeol $(1; 8 \text{ mg})$, together with eight known lupine alkaloids, $(-)$ -cytisine (2) , $(-)$ -N-methylcytisine, $(-)$ -Nformylcytisine, (-)-epibaptifoline, (-)-lusitanine, epilupinine, N-3-oxobutylcytisine and rhombifoline.

 $(+)$ -Hupeol (1) was obtained as colourless needles from CH₂Cl₂-MeOH, α _D +32.3 (c 0.263, EtOH). The molecular formula, $C_{11}H_{13}NO_3$ [Found (EIMS): m/z , 207.0882. C₁₁H₁₃NO₃ requires M_r , 207.0894], contains one nitrogen and one hydrogen less and two oxygens more than that of the typical C_{11} lupine alkaloid (-)-cytisine

 $(C_{11}H_{14}N_2O)$. The IR spectrum of 1 showed an absorption band at 3300 cm^{-1} (OH). The mass spectrum of 1 revealed prominent fragment ions at m/z 160 (43%) and 146 (97) which are characteristic of lupine alkaloids having a 2-pyridone ring such as in $2⁴$. The ¹H NMR spectrum (CD_3OD) of 1 exhibited two sets of signals in a 3:1 ratio, indicating that 1 was a 3:1 mixture of two structurally related compounds 1a and 1b, respectively, although 1 showed a single spot on TLC in several solvent systems.

The ${}^{1}H$ and ${}^{13}C$ NMR spectra of 1a and 1b (Table 1) were assigned by analysis of the $^1H-^1H$ COSY and $^1H-^{13}C$ COSY spectra. The similarity of the spectra of 1a and 1b with those of $(-)$ -cytisine (2) (Table 1) suggested that both 1a and 1b had structures very similar to that of 2. The hemiacetal structures for 1a and 1b were presumed from downfield shifts of the ${}^{1}H$ and ${}^{13}C$ signals at the 11 and 13 positions compared with those of 2. The C-9 signals of 1a and $1b$ were shifted downfield by $4-5$ ppm, compared with those of 2, while the C-7 signals were only slightly shifted, indicating that the hydroxy groups of 1a and 1b were both situated at the 11 position.

The stereochemistry of the hydroxy group of 1a and 1b was concluded to be axial (α) and equatorial (β) , respectively, by comparison of the 13 C signals at the 8, 10 and 13 positions of 1a with those of 1b. The 13 C signals of $C-8$ and $C-13$ of 1a were at a higher field than those of 1b, and the signal of $C-10$ of 1b was at a higher field than that

Table 1 ¹H and ¹³C NMR data for (+)-hupeol (1a and 1b) (CD₃OD) and (-)-cytisine (2) (CDCl₃)^a

Carbon No.	1a		1b		2	
	$\delta_{\rm C}$	δ H	$\delta_{\rm C}$	$\delta_{\rm H}$	$\delta_{\rm C}$	$\delta_{\rm H}$
2	165.7		165.7		166.6	
3	116.9	6.45 (dd, J 9.0, 1.2)	116.9	6.45 (dd, J 9.0, 1.3)	117.8	6.45 (dd, J 8.8, 1.4)
	141.5	7.49 (dd, J 9.0, 6.5)	141.5	7.49 (dd, J 9.0, 6.5)	142.1	7.29 (dd, J 8.8, 6.7)
5	107.8	6.28 (dd, J 6.5, 1.2)	107.8	6.28 (dd, J 6.5, 1.2)	108.9	5.98 (dd, J 6.8, 1.4)
6	152.8		152.8		153.4	
	36.0	2.94 (m)	36.8	2.94 (m)	36.9	2.90 (m)
8 H $_{\beta}$ H_{α}	19.9	2.52 (d, J 12.8) 1.81 (dd, J 12.8, 3.1)	25.5	2.09 (dd, J 11.9, 3.0) 2.11 (d, J 11.9)	27.3	1.96 (m) 1.96 (m)
9	34.6	2.35 (m)	33.6	2.35 (m)	29.7	2.32 (m)
10 H_B H_{γ}	49.6	4.11 (d, J 15.8) 3.85 (dd, J 15.8, 6.8)	44.1	4.47 (d, J 16.0) 3.66 (dd, J 16.0, 6.8)	51.8	4.13 (d, J 15.3) 3.89 (dd, J 15.3, 6.7)
11 H_B H_{γ}	97.2	5.12(s)	97.9	4.91 (d, J 3.0)	53.6	3.11 (m) 3.07 (m)
13 H_B H_{α}	66.8	3.43 (dd, J 10.8, 1.7) 4.37 (dd, J 10.8, 1.8)	73.3	3.82 (dd, J 11.3, 1.9) 3.93 (dd, J 11.3, 1.8)	54.6	3.03 (m) 3.09 (m)

^aJ Values in Hz.

†This is a Short Paper as defined in the Instructions for Authors, Section 5.0 [see *J. Chem. Research* (S) *, 1998*, Issue 1]; there is therefore no corresponding material in J . Chem. Research (M) . *To receive any correspondence (e-mail: ohmiya@hoshi.ac.jp).

of 1a, which could be explained by a γ effect of the hydroxy group (Scheme 1). Therefore, it was concluded that $(+)$ hupeol (1) was an inseparable equilibrium mixture (3:1) of hemiacetals 1a and 1b.

Fig. 1 A possible biosynthetic pathway for $(+)$ -hupeol (1)

It is generally accepted in the biosynthesis of lupine alkaloids that tetracyclic sparteine-type alkaloids are first produced from three units of L-lysine and then metabolized oxidatively to tricyclic cytisine-type alkaloids via tetracyclic anagyrine-type alkaloids (Fig. 1).⁴ (-)-Cytisine (2) is considered to be the ultimate metabolite in the biosynthetic pathway of the lupine alkaloids. $(+)$ -Hupeol (1) does not have a basic amino group, which is one of the important characteristics of the other alkaloids, but has a structure closely related to that of 2. Thus, $(+)$ -hupeol (1) could be regarded as an intermediate in the metabolism of lupine alkaloids to non-basic compounds. To the best of our knowledge, this is the first example of such an intermediate in the biosynthesis of lupine alkaloids.

Investigation of the absolute stereochemistry of $(+)$ hupeol (1) is currently being undertaken in our laboratories.

Experimental

Mps are not corrected. High- and low-resolution mass spectra were measured at 70 eV using a direct-inlet system. ¹H NMR (270 or 500 MHz) and ¹³C NMR (125 MHz) spectra were recorded using TMS as an internal standard.

Isolation of $(+)$ -Hupeol (1).—The crude alkaloid fraction (5.5 g) obtained from the 75% MeOH extracts was subjected to chromatography on a silica gel column (Merck, type 60 , 230-400 mesh; 410 g) with $CH_2Cl_2-MeOH-25%$ NH₄OH (43:6:1), monotoring with TLC, to give 17 fractions. The fourth fraction (25 mg), the 1-rich fraction, was separated by silica gel column chromatography with CH₂Cl₂-AcOEt-MeOH (5:5:1) to yield (+)-*hupeol* (1; 8 mg), colourless needles from CH₂Cl₂-MeOH, $[\alpha]^{\text{23}}_{\text{D}}+32.3$ (c = 0.263, EtOH); m/z (EI) 207.0882 (M⁺, C₁₁H₁₃NO₃ requires 207.0894, 57%), 190.0886 (M⁺-OH, C₁₁H₁₂NO₂ requires 190.0868, 4), 189.0801 (M⁺-H₂O, C₁₁H₁₁NO₂ requires 189.0790, 17), 178.0859 $(M⁺-CHO, C₁₀H₁₂NO₂$ requires 178.0866, 20), 160.0761
(C₁₀H₁₀NO requires 160.0761, 43), 149.0842 (C₉H₁₁NO requires 149.0841, 68), 148.0771 (C₉H₁₀NO requires 148.0761, 100), 1246.0613 (C₉H₈NO requires 146.0607, 97), 138 (38), 117 (36), 93 (35); v_{max} (KBr)/cm⁻¹ 3300 (OH), 1650 (C=O).

Received, 8th December 1997; Accepted, 8th December 1997 Paper E/7/08797G

References

- 1 K. Saito, T. Yoshino, T. Sekine, S. Ohmiya, H. Kubo, H. Otomasu and I. Murakoshi, Phytochemistry, 1989, 28, 2533.
- 2 S. Ohmiya, H. Kubo, H. Otomasu, K. Saito and I. Murakoshi, Heterocycles, 1990, 30, 537.
- 3 H. Kubo, S. Ohmiya and I. Murakoshi, Can. J. Chem., 1994, 72, 214; S. Ohmiya, H. Kubo, Y. Nakaaze, K. Saito, I. Murakoshi and H. Otomasu, Chem. Pharm. Bull., 1990, 39, 1123.
- 4 S. Ohmiya, K. Saito and I. Murakoshi, in The Alkaloids, ed. G. A. Cordell, Academic Press, New York, 1955, vol. 47, p 1.